Omphacite-diopside vein in an omphacitite block from the Osayama serpentinite melange, Sangun-Renge metamorphic belt, southwestern Japan

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Abstract

Omphacite $(Jd_{46.1-52.0}Ae_{0-8.4}Aug_{48.0-51.2})$ and diopside $(Jd_{4.3-6.3}Ae_{0-0.4}Aug_{93.6-95.6})$ coexist in a vein cutting an omphacitite block in a serpentinite melange of the Oeyama ophiolite, central Chugoku Mountains. The compositional gap between omphacite and diopside is significantly wider than for other omphacite-diopside pairs reported in the literature. The intergrowth texture of the omphacite-diopside vein suggests that the clinopyroxene pair was contemporaneously crystallized in the fracture from a Ca-bearing, alkali-rich fluid in a sub-solvus condition. Such a fluid may have been introduced from the surrounding serpentinized clinopyroxene-bearing harzburgite. The stability of omphacite and Al-rich pumpellyite in the matrix and the omphacite-diopside vein indicate that the earlier matrix recrystallization and later fracture filling may have both taken place under high-P-T condition during the melange-forming stage.

KEYWORDS: omphacite, diopside, immiscibility, serpentinite melange, Oeyama ophiolite, Sangun-Renge metamorphic belt.

Introduction

OMPHACITE is a common metamorphic mineral in blueschist and eclogite facies metamorphic rocks, and also in metasomatic sodic pyroxene-rich rocks such as jadeitite and omphacitite. Coexisting omphacite-diopside pairs have been reported from high-pressure metamorphic rocks by several authors (e.g. Brown *et al.*, 1978; Carpenter, 1980*a*; Enami and Tokonami, 1984), and they provide some evidence of a miscibility gap in the omphacite-diopside system. Coexisting omphacite and diopside have now been found in a vein from an omphacite block in a serpentinite melange of the Oeyama ophiolite, central Chugoku Mountains. As described in this paper, the compositional gap is wider than the other omphacite-diopside pairs reported in the literature.

Geology of the Osayama serpentinite melange

The Osayama serpentinite melange is located in the Sangun-Renge metamorphic belt of the Chugoku Mountains, in the Inner Zone of southwestern Japan. The Sangun-Renge metamorphic belt is the Carboniferous high-P-T type metamorphic belt,

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which tectonically underlies the Early Palaeozoic Oeyama ophiolite (e.g. Ishiwatari, 1991), and may be a western counterpart of the Omi glaucophane schist (Banno, 1958) in the Hida Mountains.

The Osayama serpentinite melange is developed beneath the Osayama peridotite body of the Oeyama ophiolite in the central Chugoku Mountains, as a fragment of the Oeyama ophiolite (Fig. 1). It is a typical tectonic melange with a serpentinite matrix, which has probably been developed along a thrust fault between the Ordovician Oeyama ophiolite and Carboniferous Sangun-Renge blueschist nappe during the emplacement of the Oeyama ophiolite. The melange overlying the massive peridotite body is tectonically underlain by the 180 Ma basic schist of the Sangun-Chizu belt, which is in contact with the unmetamorphosed shallow marine sediments of the Jurassic Yamaoku Formation on the north because of a high-angle fault. All these rocks are unconformably overlain by the early Cretaceous Kyomiyama conglomerate. The massive peridotite body and the Sangun-Chizu schist are affected by contact metamorphism of the Cretaceous granitic intrusion on the west (Nozaka and Shibata, 1995). The protolith of the serpentinite matrix is harzburgite with minor dunite



FIG. 1. Geological map of the Osayama serpentinite melange. [LP: schist of lawsonite-pumpellyite zone; E: schist of epidote zone; Gb: gabbro; Dl: dolerite; At: albitite]. The omphacitite block described in this paper occurs in the western part of the melange.

(Nozaka and Shibata, 1994; Tsujimori, in press), which may have been the residual peridotite of the Oeyama ophiolite.

The blueschist-facies schists, fragments of the Oevama ophiolite (serpentinized peridotite, gabbro, dolerite) and metasomatic rocks (albitite, jadeitite, omphacitite, tremolite schist, etc.) are enclosed as tectonic blocks of various size (10 cm to 1.5 km in length) in a serpentinite matrix (Tsujimori, in press). The blueschist-facies schists are divided into a lawsonite-pumpellyite zone (230-270°C at 6-8 kbar) and an epidote zone by the mineral assemblages of the mafic blueschists. The epidote zone blocks contain two varieties i.e. a garnet-free, lower grade block (>300°C at 6-8 kbar) and a garnetbearing higher-grade block (garnet-glaucophane schist) (400-530°C at 11-13 kbar). The gabbro and dolerite blocks also contain blueschist-facies mineral assemblages corresponding with the lawsonite-pumpellyite zone. Tsujimori (1995a) found relics of the eclogite-facies mineral assemblage (620°C at 15 kbar) in a garnet-glaucophane schist block. Tsujimori (1995b) reported a chromian omphacite-bearing tremolite schist of peridotite origin, suggesting that a part of the Oeyama ophiolite has experienced a high-P-T metamorphism with the Sangun-Renge metamorphic rocks. The blueschist blocks give phengite K-Ar ages of around 320 Ma (Tsujimori and Itaya, submitted).

Petrography

The omphacitite block, 1×1 m in size, occurs at a small outcrop in the central part of the melange. This is a well foliated, dark-green to black rock. It is surrounded by sheared serpentinite.

The omphacite-diopside veins (< 0.4 mm wide) consist mainly of omphacite and diopside with minor amounts of pumpellyite and biotite. The veins cut the foliation at high angles (Fig. 2a). Omphacite in the vein occurs as radiating anhedral crystals which are intergrown with fan-shaped diopside crystal (< 0.3 mm) (Fig. 2b). Omphacite shows weak pleochroism between X' = colourless and Z' = faint green, and exhibits lower birefringence than the coexisting diopside. Diopside in the vein has a radiating habit and tends to be oriented perpendicular to the vein wall. The extinction positions of the two adjacent clinopyroxenes are different from each other. The two-pyroxene pair may have grown from the vein wall, and the texture suggests that the omphacite and diopside have contemporaneously crystallized in the vein.

The omphacitite matrix consists mainly of omphacite with small amounts of chlorite, pumpellyite and titanite. Albite, apatite and biotite occur as accessories in the matrix, and quartz is absent.



FIG. 2. (a) Back-scattered electron image of the occurrence of the omphacite-diopside vein cutting a foliated omphacitite matrix. (b) Back-scattered electron image of coexisting omphacite (Omp) and diopside (Di) of the omphacite-diopside vein.

Titanite and apatite are not found in the omphacitediopside vein. The albite-rich layer (< 3 mm wide) is rarely observed in the matrix. The foliation is defined by preferred orientation of elongated omphacite crystals. Matrix omphacite occurs as prismatic crystals (0.05-0.2 mm in length), and shows weak pleochroism between X' = colourless and Z' = pale green. A small number of crystals have a thin green rim. Chlorite occurs as oriented, tabular, colourless crystals, and shows a grey or brown interference colour. The elongation is positive. Pumpellyite is colourless and occurs as prismatic crystals (< 0.05 mm in length). Apatite occurs as anhedral crystals (< 0.1 mm).

Mineral chemistry

Chemical analyses of minerals were carried out with an AKASHI ALPHA-30A scanning electron micro-

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| | Omphacite-diopside vein | | | | | | Matrix | | | |
|---|-------------------------|--------|--------|--------|----------|--------|--------|-----------|--------|--------|
| | Omphacite | | | | Diopside | | | Omphacite | | |
| wt.% | | 1 | | | 1 | | | 1 | | |
| SiO ₂ | 56.79 | 56.50 | 55.86 | 52.72 | 53.66 | 53.05 | 56.76 | 55.58 | 55.98 | 51.48 |
| TiO ₂ | 0.45 | 0.20 | 0.29 | 0.15 | 0.06 | 0.01 | 0.19 | 0.23 | 0.11 | 0.14 |
| $A1_2O_3$ | 10.51 | 11.54 | 11.00 | 2.12 | 1.43 | 1.20 | 9.73 | 9.64 | 8.15 | 2.53 |
| Cr_2O_3 | 0.03 | 0.15 | 0.16 | 0.12 | 0.17 | 0.15 | 0.20 | 0.22 | 0.12 | 0.11 |
| FeO* | 7.13 | 7.06 | 7.51 | 10.57 | 10.88 | 10.33 | 6.09 | 5.77 | 7.34 | 19.53 |
| MnO | 0.13 | 0.16 | 0.19 | 0.33 | 0.40 | 0.34 | 0.11 | 0.04 | 0.22 | 0.41 |
| MgO | 6.12 | 5.39 | 7.00 | 11.13 | 10.07 | 11.06 | 7.06 | 7.79 | 7.38 | 4.11 |
| CaO | 11.55 | 11.71 | 11.38 | 21.79 | 22.96 | 23.06 | 13.76 | 14.16 | 14.60 | 18.15 |
| Na ₂ O | 7.46 | 7.31 | 7.04 | 1.34 | 0.79 | 0.79 | 6.35 | 6.74 | 6.21 | 3.91 |
| K ₂ O | 0.03 | 0.03 | 0.03 | 0.09 | 0.00 | 0.03 | 0.00 | 0.06 | 0.00 | 0.00 |
| Total | 100.20 | 100.05 | 100.46 | 100.34 | 100.42 | 100.02 | 100.25 | 00.23 | 100.11 | 100.37 |
| Number of cations on the basis of 6 oxygens | | | | | | | | | | |
| Si | 2.032 | 2.022 | 1.998 | 1.977 | 2.013 | 1.998 | 2.029 | 1.997 | 2.028 | 2.008 |
| Ti | 0.012 | 0.005 | 0.008 | 0.004 | 0.002 | 0.000 | 0.005 | 0.006 | 0.003 | 0.004 |
| Al | 0.443 | 0.487 | 0.464 | 0.094 | 0.063 | 0.053 | 0.410 | 0.408 | 0.348 | 0.116 |
| Cr | 0.001 | 0.004 | 0.004 | 0.004 | 0.005 | 0.004 | 0.006 | 0.006 | 0.004 | 0.003 |
| Fe* | 0.213 | 0.211 | 0.225 | 0.331 | 0.341 | 0.325 | 0.182 | 0.173 | 0.222 | 0.637 |
| Mn | 0.004 | 0.005 | 0.006 | 0.010 | 0.013 | 0.011 | 0.003 | 0.001 | 0.007 | 0.014 |
| Mg | 0.326 | 0.288 | 0.374 | 0.622 | 0.563 | 0.621 | 0.376 | 0.417 | 0.399 | 0.239 |
| Ca | 0.443 | 0.449 | 0.436 | 0.876 | 0.923 | 0.930 | 0.527 | 0.545 | 0.567 | 0.759 |
| Na | 0.517 | 0.507 | 0.488 | 0.097 | 0.057 | 0.058 | 0.440 | 0.470 | 0.436 | 0.296 |
| Κ | 0.001 | 0.001 | 0.001 | 0.004 | 0.000 | 0.001 | 0.000 | 0.003 | 0.000 | 0.000 |
| Total | 3.993 | 3.981 | 4.004 | 4.020 | 3.980 | 4.003 | 3.978 | 4.026 | 4.012 | 4.076 |

TABLE 1. Representative microprobe analyses of clinopyroxenes in the omphacitite

* Total Fe.

scope equipped with a Philips EDAX-9100 energydispersive analytical system at Kanazawa University. Natural and synthetic minerals such as periclase, hematite, wollastonite, albite and chromite were used as standards, and a program based on Bence and Albees (1968) method and written by A. Ishiwatari was used for data processing. The analyses were done at 20 kV accelerating voltage, 1 nA sample current on MgO and <10 μ m beam diameter. Representative chemical analyses of clinopyroxenes in omphacitite are listed in Table 1.

Clinopyroxenes

The nomenclature of sodic pyroxene is based on the classification of Morimoto *et al.* (1988). The estimation of Fe^{2+} and Fe^{3+} was based on the assumption of four total cations and six oxygens. Jadeite, aegirine and augite (diopside) components of the metamorphic pyroxene were calculated as follows: (1) Jd = AI^{VI} ; (2) Ae = Fe^{3+} ; (3) Di(Aug)

= Ca, (4) and these components were normalized to 100. Analysed clinopyroxenes are plotted in a jadeite–aegirine–diopside ternary diagram (Fig. 3).

Omphacite in the omphacite-diopside vein has a significantly higher jadeite component $(Jd_{46,1-52,0}Ae_{0-8,4}Aug_{48,0-51,2})$ than the omphacite in the matrix $(Jd_{37,1-46,3}Ae_{0-5,3}Aug_{53,7-60,4})$. The FeO (total iron) content (av. 7.4 wt.%) of the vein omphacite is slightly higher than that of the matrix omphacite (av. 6.6 wt.%). Diopside coexisting with omphacite in the vein is poor in the jadeite component $(Jd_{4,3-6,3}Ae_{0-0,4}Aug_{93,6-95,6})$. The green rim on matrix omphacite is also poor in jadeite, but has a slightly higher aegirine component than the vein diopside $(Jd_{4,2-11,3}Ae_{11,0-16,1}Aug_{73,9-82,0})$.

Some jadeitites in the Osayama melange also contain omphacite. However, the omphacite in the jadeitite shows quite different occurrences. The omphacite $(Jd_{42.5-71.3}Ae_{0-5.1}Aug_{28.7-57.5})$ occurs as overgrowths on pure $(Jd_{>98})$ or impure (Jd_{84-96}) jadeite, and as veinlet clots (2-5 mm wide), consisting of radial aggregates of needle-like crystals.



FIG. 3. Chemical compositions of the clinopyroxenes in the omphacitite block from the Osayama serpentinite melange plotted on a jadeite (Jd)–aegirine (Ae)–diopside (Di) ternary diagram. The compositional ranges reported by Brown *et al.* (1978), Carpenter (1980*a*) and Enami and Tokonami (1984) are also shown in small diagrams.

The omphacite in jadeitite may have been retrogressively formed in jadeitite.

Other minerals

Pumpellyite in omphacitite has an Al-rich composition with Al/(Al+Fe*+Mg) = 0.78-0.81. The Fe*/(Fe*+Mg) ratio (0.28-0.39) is slightly lower than that of pumpellyite from other blueschist blocks in the Osayama serpentinite melange (0.41-0.69). Chlorite has a composition of Fe*/(Fe*+Mg) = 0.40 and Al/(Al+Fe*+Mg) = 0.33. Biotite in omphacitite has a composition of Mg/(Mg+Fe²⁺) = 0.62, TiO₂ = 0.4 wt.% and Si = 5.8 (p.f.u. for O = 22).

Discussion

Comparison with the other omphacite-diopside pairs

The occurrence of coexisting omphacite and diopsidic pyroxene pairs has previously been reported from high-P metamorphic rocks. The coexistence is often ascribed to the presence of a miscibility gap between omphacite and diopsidic

pyroxene. For example, omphacite and diopside coexist in a blueschist specimens from Piemont, Italy $(Jd_{35-50} \text{ and } Jd_{3-12}$: Brown *et al.*, 1978) and from the Russian River, Franciscan Complex, U.S.A. $(Jd_{28-41} \text{ and } Jd_{8-15}: Carpenter, 1980a)$. The paired clinopyroxenes reported by Brown et al. (1978) occur in an unusual manganese-rich, metamorphic rock containing manganoan minerals such as braunite. hollandite, piemontite, manganoan phengite and manganoan phlogopite. The omphacite coexisting with diopside contains up to 2.5 wt.% MnO and less than about 5 mole% aegirine component. Peak metamorphic condition was estimated as 8 + 1kbar and 300 \pm 50°C. The paired clinopyroxenes reported by Carpenter (1980a) form sector-zoned composite grains constituting emerald-green veins (up to 1 cm wide) and cutting an epidote-blueschist. The veins rarely contain quartz and white mica. The omphacite has about 25 mole% aegirine component.

Several studies on the microstructure of sodic pyroxene (e.g. Matsumoto *et al.*, 1975) have revealed that cation ordering can occur in omphacite with compositions close to $Jd_{50}Di_{50}$, causing a symmetry change from *C*-face-centred (*C*2/*c*) to primitive (*P*2/*n*). Carpenter (1979) determined the compositional limits of the ordered P2/n omphacite relative to disordered C2/c pyroxenes in blueschist-facies rocks. Recently, Carpenter *et al.* (1990) thermodynamically described the cation ordering in omphacite using Landau theory.

Two different possible equibrium phase relations between omphacite and diopside were presented. Carpenter (1980b) suggested the phase relations in the jadeite-diopside system shown in Fig. 4a. The apparent consolute temperature for the solvus between P2/n omphacite and C2/c diopside is about 500°C. He proposed a second-order transformation mechanism for the $C2/c \rightleftharpoons P2/n$ transition. On the other hand, Enami and Tokonami (1984) found two different coexisting clinopyroxene pairs, C2/c omphacite $(Jd_{35}) - C2/c$ diopside (Jd_{15}) in the epidote-amphibolite facies stage (above 500C), and a P2/n omphacite (Jd₄₄)-C2/c diopside (Jd₁₀) in later epidote-blueschist facies stage (below 500°C) from the Iratsu eclogitic amphibolite mass, Sambagawa belt, Japan. They proposed a first-order transformation model for the transformation C2/c to P2/n to explain the persistence of the C2/c omphacite structure. Although the two interpretations for



FIG. 4. (a) Phase relations in the Fe³⁺-free diopside–jadeite system proposed by Carpenter (1980b). See
Fig. 3 for abbreviations. (b) Phase relations between
Fe³⁺-free diopside and omphacite proposed by Enami and Tokonami (1984), which were partly based on Carpenter's diagram. BS: blueschist-facies, EA: epidote-amphibolite facies.

omphacite transformation from P2/n to C2/c are quite different, the solvus between P2/n omphacite and C2/c diopside becomes wider with decreasing temperature in both cases. Although the space group of omphacite coexisting with diopside in the vein from the Osayama is not yet determined, it is probably ordered (P2/n), judged from its high jadeite component and low temperature of formation.

The compositional gap between the omphacite and diopside in the Osayama vein is significantly wider than for other studies (Fig. 3). This may be due to the lower temperature of formation. The presence of Alrich pumpellyite and the absence of epidote in the Osayama omphacitite constrains the metamorphic temperature condition to <350°C at 6-8 kbar. This temperature is lower than for the other omphacitediopside pairs. As mentioned above, the coexisting omphacite and diopside in the Osayama omphacitite have a radiating habit. This texture suggests that the omphacite and diopside underwent a relatively rapid rate of growth. The rapid growth would probably have resulted in omphacite with a degree of cation order, less than the case of equilibrium growth at that temperature. In general, the miscibility gap is highly sensitive to cation ordering. Although the omphacitediopside pair in the Osayama serpentinite melange may not have achieved strict thermodynamic equilibrium, the simultaneous growth of the two clinopyroxenes with a wide compositional gap suggests the lowest temperature (< 350°C) ever reported. The low temperature has undoubtedly promoted cation ordering.

Geological significance of the omphacite-diopside vein

Omphacite veins have been found at many localities in the Franciscan metamorphic rocks, such as in blueschist, amphibolite, greenstone and metagreywacke. Essene and Fyfe (1967) described some omphacite veins cutting blueschists from the Franciscan Complex. The vein omphacite occurs as prismatic crystals normal to the vein wall, or as radial aggregates, and Essene and Fyfe interpreted these textures as having been formed in a hydrostatic environment during omphacite crystallization. Coleman (1961) reported jadeite veins cutting blueschist in the Franciscan Complex, New Idria, U.S.A., and he interpreted the jadeite vein as having been produced by mobilization of a liquid rich in the jadeite component and low in H₂O during tectonic movement of the host blueschist. Philippot and Kienast (1989) described an omphacite-rich vein that cut mylonite zones in an eclogite-facies metagabbro, Monviso, Western Alps, Italy, and they found sector-zoned omphacite-chloromelanite

The texture of our omphacite-diopside vein suggests that the clinopyroxene pair was formed contemporaneously in the fracturing stage following the metasomatism from a Ca- and alkali-rich fluid at a lower temperature than the solvus between omphacite and diopside. Such a fluid might be introduced from the surrounding serpentinite, because CaO may have been released from clinopyroxene-bearing harzburgite during its serpentinization. The stability of omphacite and Al-rich pumpellvite, both in the matrix and in the vein. indicates that the matrix-forming stage and the later fracturing stage may have been under relatively highpressure conditions of 6-8 kbar at $< 350^{\circ}$ C. The omphacite-diopside vein may have formed during the melange-forming stage at relatively high-P-T.

Conclusion

The compositional gap between the coexisting omphacite (Jd_{46-52}) - diopside (Jd_{4-6}) in the Osayama omphacite vein is significantly wider than that for other omphacite-diopside pairs reported in the literature. This may be due to the lower temperature of their formation. They may have grown simultaneously from a vein-filling fluid in the omphacitite block at relatively high pressure during the melange-forming stage.

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References

- Bence, A.E. and Albee, A.L. (1968) Empirical correction factors of the electron microanalysis of silicates and oxides. *J. Geol.*, **76**, 382–403.
- Banno, S. (1958) Glaucophane schists and associated rocks in the Omi district, Japan. J. Geol. Geogr. Japan, 29, 29–44.
- Brown, P., Essene, E.J. and Peacor, D.R. (1978) The mineralogy and petrology of manganese-rich rocks

from St. Marcel, Piedmont, Italy. *Contrib. Mineral. Petrol.*, **67**, 227–32.

- Carpenter, M.A. (1979) Omphacite from Greece, Turkey, and Guatemala: composition limits of cation ordering. *Amer. Mineral.*, **65**, 313–20.
- Carpenter, M.A. (1980a) Composition and cation order variations in a sector-zoned blueschist pyroxene. *Amer. Mineral.*, 65, 313–20.
- Carpenter, M.A. (1980b) Mechanisms of exsolution in sodic pyroxenes. *Contrib. Mineral. Petrol.*, 71, 289–300.
- Carpenter, M.A., Domeneghetti, M.-C. and Tazzoli, V. (1990) Application of Landau theory to cation ordering in omphacite I: Equilibrium behavior. *Eur. J. Mineral.*, 2, 7–18.
- Coleman, R.G. (1961) Jadeite deposits of the Clear Creek area, New Idria district, San Benito County, California. J. Petrol., **2**, 209–47.
- Enami, M. and Tokonami, M. (1984) Coexisting sodic diopside and omphacite in a Sanbagawa metamorphic rocks, Japan. *Contrib. Mineral. Petrol.*, 86, 241-7.
- Essene, E.J. and Fyfe, S. (1967) Omphacite in Californian metamorphic rocks. *Contrib. Mineral. Petrol.*, **15**, 1–23.
- Ishiwatari, A. (1991) Time-space distribution and petrologic diversity of Japanese ophiolites. In: *Ophiolite Genesis and Evolution of the Oceanic Lithosphere* (Tj. Peters *et al.*, eds.) Kluwer Academic Publishing, Amsterdam, 723–43.
- Matsumoto, T., Tokonami, M. and Morimoto, N. (1975) The crystal structure of omphacite. *Amer. Mineral.*, **60**, 634–41.
- Morimoto, N., Fabries, J., Ferguson, A.K., Ginzburg, I.V., Ross, M., Seifert, F.A., Zussman, J., Aoki, K. and Gottardi, G. (1988) Nomenclature of pyroxenes. *Amer. Mineral.*, **73**, 1123–33.
- Nozaka, T. and Shibata, T. (1994) Petrography of primary peridotites from the Ohsa-yama area, Okayama Prefecture. *Okayama Univ. Earth Sci. Report*, **1**, 1–8.
- Nozaka, T. and Shibata, T. (1995) Mineral paragenesis in thermally metamorphosed serpentinites, Ohsayama, Okayama Prefecture. *Okayama Univ. Earth Sci. Report*, **2**, 1–12.
- Philippot, P. and Kienast, J. R. (1989) Chemicalmicrostructural changes in eclogites in eclogitefacies shear zones (Monviso, Western Alps, north Italy) as indicators of strain history and the mechanism and scale of mass transfer. *Lithos*, 23, 179–200.
- Tsujimori, T. (1995*a*) Finding of eclogite from the Sangun-Renge metamorphic belt of the Chugoku district, the Inner Zone of southwestern Japan. *Abstracts of 1994 Annual Meeting of the Japanese Association of Mineral., Petrol. Econ. Geol.* p. 50. (Japanese with English abstract).

- Tsujimori, T. (1995b) Chromian omphacite and chromian diopside in the tremolite schist blocks from the Osayama serpentinite melange, Sangun-Renge metamorphic belt, southwestern Japan. *Abstracts of 1995 Joint Annual Meeting of Janan. Assoc. Min. Petr. Econ. Geol., Min. Soc. Japan and Soc. Resource Geol.* p. 33. (in Japanese).
- Tsujimori, T. (in press) Geology of the Osayama serpentinite melange in the central Chugoku Mountains, southwestern Japan: 320 Ma blueschistbearing serpentinite melange beneath a peridotite

body of the Oeyama ophiolite. J. Geol. Soc. Japan. (Japanese with English abstract).

Tsujimori, T. and Itaya, T. (submitted) Blueschist-facies metamorphism during Carboniferous orogeny in southwestern Japan: phengite K-Ar ages of blueschist-facies tectonic blocks in a serpentinite melange beneath Early Paleozoic Oeyama ophiolite. *Tectonophys*.

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